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COMPARISON BETWEEN PARTICLE SIZE DISTRIBUTION OF HEXOGEN AND OCTOGEN DURING FORCED CRYSTALLIZATION

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ABSTRACT

Advanced solid rocket propellants with a low-pressure exponent and minimum signature comprise high-energy nitramine compounds such as hexogen and octogen. To achieve the desired processing conditions and increase the solid loading, the nitramine explosives are required with different grain size. The control of growth rate of crystals of nitramines by stirring, cooling and by crystal habit modifiers has been studied. The influence of these factors on crystal size distributions has been investigated and discussed. The results of both hexogen and octogen are compared.

KEY WORDS

Rocket propellants, Composite rocket propellants, Nitramine propellants, Hexogen, Octogen.

INTRODUCTION

Low vulnerability propellants (LOVA) and plastic bonded explosives (PBX) comprise a crystalline nitramines such as hexogen (RDX) or octogen (HMX). These are used to enhance performance, improve design characteristics and minimize signature. Nitramines also increase long time thermal stability, minimize thermal insulation problems in rockets and decrease pollution in the field [1-2].

The propellant must have rheological properties that permit flow into all parts of the motor case during casting [2]. If the flow is not adequate, voids or other defects can develop, causing an increase in the burning surface and may lead to detonation.

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To obtain good rheological properties, and to increase the solid loading in propellants and PBXs, especial grade RDX or HMX is needed [3].

Fine powders are usually prepared by grinding, spray drying, or by recrystallization. Grinding of nitramines is very dangerous due to its relatively high sensitivity to impact and friction [4], while spray drying towers are difficult to be designed and operated to produce such fine powders. Thus controlled recrystallization of commercially available hexogen and octogen is much more reasonable [5-7]. For obtaining fine hexogen and octogen crystals which are essential for improving the rheological properties and increasing the solid loading of nitramine propellants, three recrystallization methods are described.

MATERIALS AND METHODS

Materials:-

The materials and their sources are tabulated in Table 1.

Table 1. List of materials

Material	Trade name	Sources
Hexogen	RDX	ROF. UK.
Octogen	HMX	ROF. UK.
Acetone	CH ₃ COCH ₃	El Nasr Co. Egypt.
Polysorbate 80	Tween-80	Prolabo. France.
Dimethylsulphoxide	DMS	Aldrich Chem. Co. England

Sieving process:-

The distribution of particle size greater than 40 micron is determined by classifying it on a set of sieves placed on a shaking device (Retsch laboratory sieving machine model VS 1000).

The particles with size less than 40 micron and the average particle size (APS) were determined by using Fisher subsieve sizer.

Experimental techniques:-

The fine hexogen (RDX) and octogen (HMX) obtained experimentally by recrystallization of the commercially available RDX and HMX, following three different methods of crystallization.

1- Dissolution of nitramines in solvents:-

54 grams of RDX are dissolved in 100ml of acetone, while 68 grams of HMX are dissolved in 100ml of dimethylsulphoxide (DMS) at 60°C with gentle agitation using magnetic stirrer.

2- Control of growth by stirring and cooling of supersaturated solutions:-

The particle size is controlled through vigorous stirring of dissolved nitramines

at (200, 300,500, 600 ,800 r.p.m.), or by rapid cooling and all the other conditions affecting recrystallization process were kept constant for all cases,.Only natural cooling was used during stirring. Cooling recrystallization corresponding to rates of cooling equal to 2,3,5,6 and 8deg./min. was done.

3- Special grade nitramine from dilution crystallizer:

The dissolved hexogen / octogen is kept without stirring and on natural cooling for 24 hours, evaporation of solvent took place and the solution is supersaturated. This supersaturated was then poured into 200 ml of water at 25°C. The precipitated crystals were filtered, washed and dried.

4- Controlling growth rate of crystals by using crystal habit modifiers:

10ppm of (Tween-80) were added to the dissolved solutions of RDX and HMX and agitated gently . The effect of variation agitation time (2,3,5,6 and 8 hours) on recrystallized nitramine was studied by analysing the resulting crystals.

RESULTS AND DISCUSSION:-

The results of the effect of changing the stirring speed at 200,300,500,600 and 800 r.p.m. on the percentage of particles less than 40 micron (cumulative under size) of hexogen and octogen are tabulated in Table 2. And shown in Fig. 1.

Table 2.Effect of stirring speed on percentage of nitramine crystals (particle size is less than 40 micron

Stirring speed (R.P.M.)	Weight percentage of crystals having particle size less than 40 micron	
	Hexogen	Octogen
200	31	35
300	36	40
500	45	52
600	51	58
800	65	68

It is obvious that increasing the stirring speed, the particle size of the resulting crystals decreases and thus the percentage of crystals having particle size less than 40 micron is increased.

The results showing the effect of cooling rate (2/3//5/6/8deg./min.) on particle size distribution are tabulated in Table 3. and shown in Fig. 2. From these results one conclude that the percentage of fine nitramine is directly proportional to the rate of cooling.

Table 3. Effect of cooling rate on percentage of nitramine crystals (particle size is less than 40 micron).

Cooling rate (deg./min.)	Weight percentage of crystals having particle size less than 40 micron	
	Hexogen	octogen
2	34	42
3	37	46
5	44	55
6	47	58
8	52	67

Table 4. and Fig.3 show the particle size distribution of crystals resulting from dilution crystallization of both hexogen and octogen.

Table 4. Cumulative particle size distribution for RDX and HMX recrystallized from dilution crystallizer

D, microns	Cumulative particle size distribution (%)			
	RDX	HMX	RDX with 10% CHM	HMX with 10% CHM
300	95	98	97	98
250	93	98	96	98
210	84	90	95	98
150	57	75	85	94
125	43	63	72	85
75	20	36	50	70
50	12	21	36	55
45	10	18	32	50
30	-	10	11	40

By the process of dilution crystallization one can obtain fine crystals (less than 40 micron) of high purity. Analysis of the fine crystals (less than 40 micron) is shown in Table 4., from which it is clear that fine crystals could be obtained in the range of 5 to 30 micron by dilution crystallization.

Table 5. Cumulative under size analysis for hexogen Recrystallized from dilution crystallizer

D, micron	Cumulative under size
28	98
26	95
24	91
22	87
20	81
18	74
16	65
14	55
12	42
10	27
8	15
6	7
4	2

Table 6. shows the effect of addition of 10ppm Tween-80 as crystal habit modifier (CHM) on percentage of particles having less than 40microns diameter. Fig. 4 . Shows the influence of changing (Tween-80) time of agitation on particle size distribution. From these figures it is clear that the percentage of fine crystals increases with the increase of agitation time. The decrease of particle size is proportional to the increase of agitation time of CHM due to efficient spreading of CHM on crystal surface and. To obtain nitramine with suitable particle size distribution, it is necessary to optimize the time of agitation and the proportion of CHM.

Table 6. Effect of stirring time on percentage of nitramine crystals (particle size less than 40 microns) in the presence of CHM.

Stirring time (hours)	Weight percentage of crystals having particle size less than 40 microns	
	Hexogen	Octogen
2	30	36
3	41	50
4	51	61
6	64	85
8	73	95

CONCLUSIONS

Increasing the stirring speed and cooling rate increase considerably the formed fine nitramine crystals (less than 40 micron). Also by the process of dilution crystallization, fine crystals of high purity are obtained. Addition of crystal habit modifiers has a significant effect on particle size of produced crystals.

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